

# **Microgravity Studies of Liquid-Liquid Phase Transitions in Undercooled Alumina-Yttria Melts**

NASA Contract number NAS8-98092

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Project monitors: Pat Doty and Buddy Guynes

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*Containerless Research, Inc.*

NASA Mat. Sci. Workshop,  
Huntsville, AL, 6/26/02

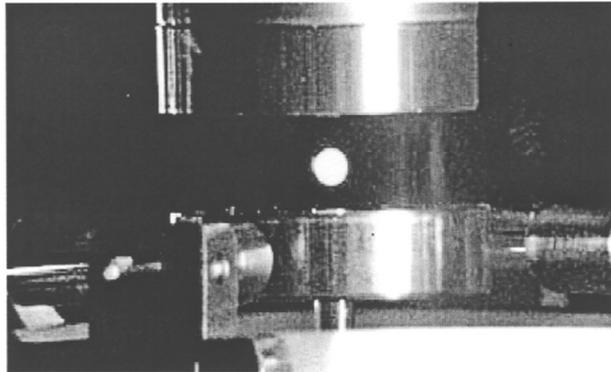
# Collaborators

- Jan Rogers, Bob Hyers, Tom Rathz, Mike Robinson, Greg Jerman, NASA Marshall Space Flight Center
- Chandra Ray, Delbert Day, U. Missouri-Rolla
- Chris Benmore, Joan Siewenie and Jacob Urquidi, Argonne National Laboratory
- Juro Majzlan, Alex Navrotsky, UC-Davis
- Brian Phillips, SUNY Stony Brook
- P-F. Paradis, T. Ishikawa, J. Yu, S. Yoda, NASDA Tsukuba Research Center

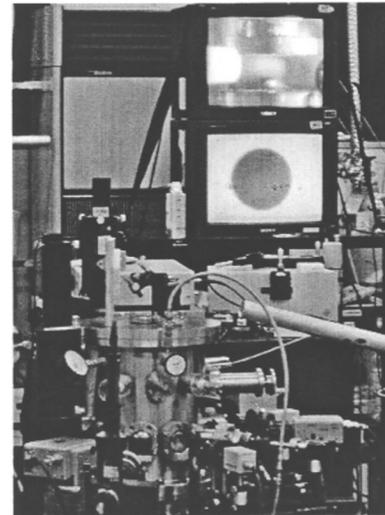
# Outline

- Background
  - techniques
  - undercooling, glass formation, applications
  - role of low gravity experiments
- Current activities
  - melt structure
  - glass structure
  - melt properties
- Future/ongoing activities
  - kinetics of L-L transition
  - measurements of liquid properties
  - detailed structure model

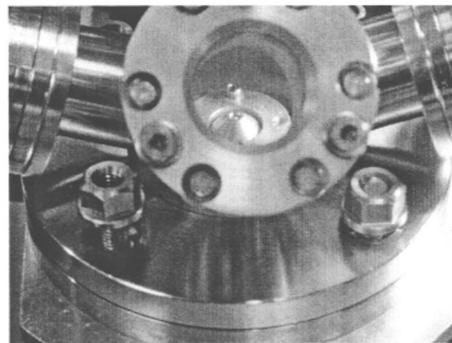
Containerless techniques used to avoid contamination of corrosive melts, control melt chemistry, and access highly non-equilibrium oxide liquids



**Electrostatic Levitation (ESL)**  
at NASA Marshall Space  
Flight Center [RSI, 64, 2961 (1993)]



**Pressurized ESL** at  
NASDA Tsukuba  
Space Center [RSI,  
72, 2811 (2001)]



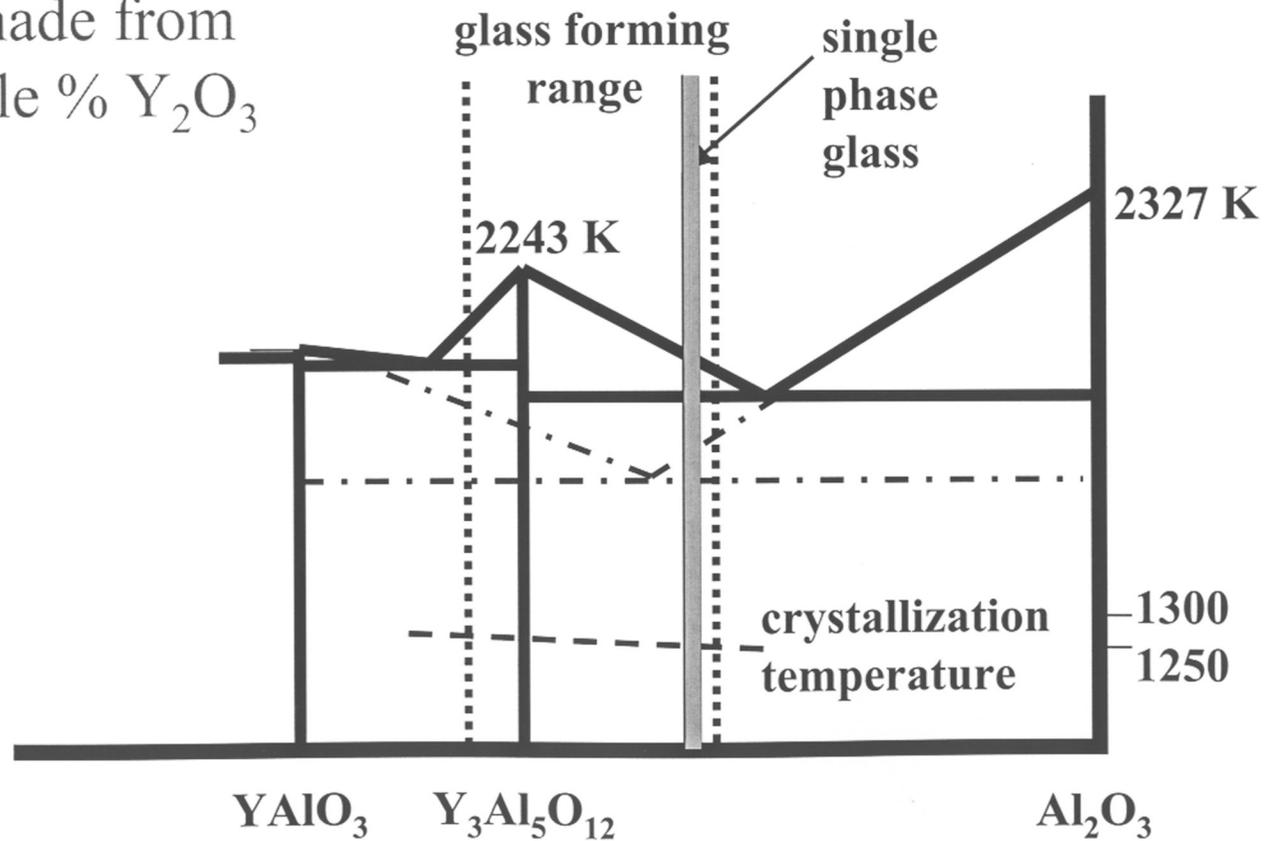
**Aerodynamic Levitation**  
[Microg. Sci. Tech., 7, 279 (1995)]



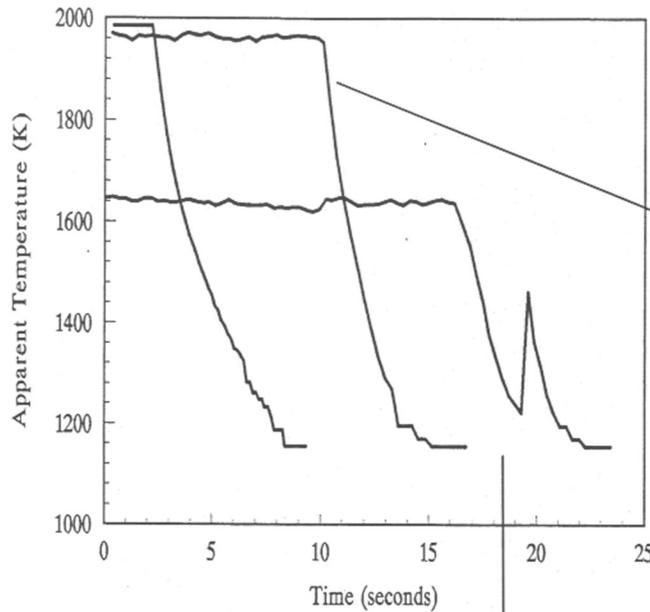
**Aero-acoustic Levitation**  
[Science, 287 (2000),  
RSI, 65, 456 (1994)]

# Compositions under investigation

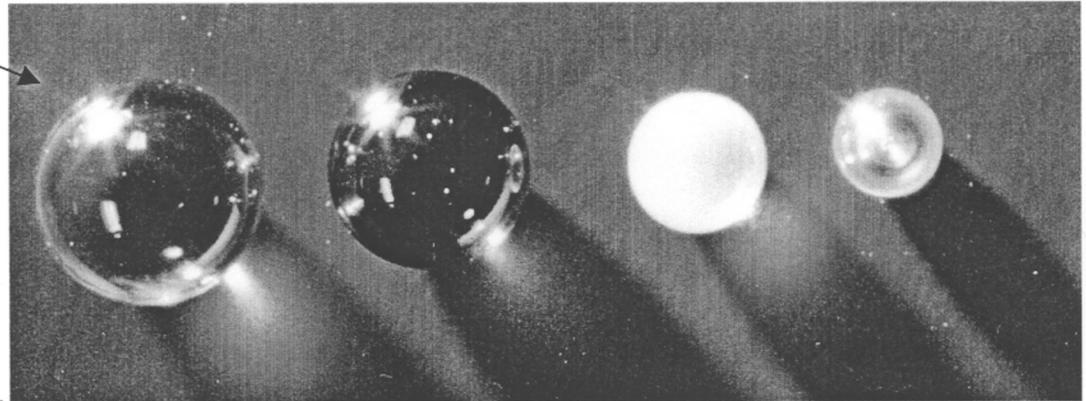
Glasses made from  
23-42 mole %  $Y_2O_3$



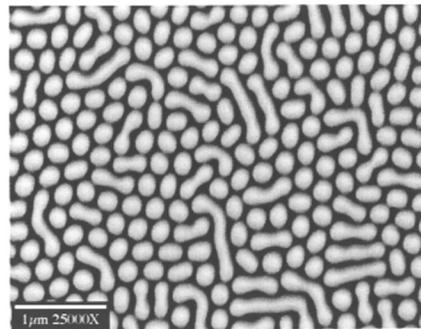
# Glasses made from RE-Al oxide materials



3.5 mm

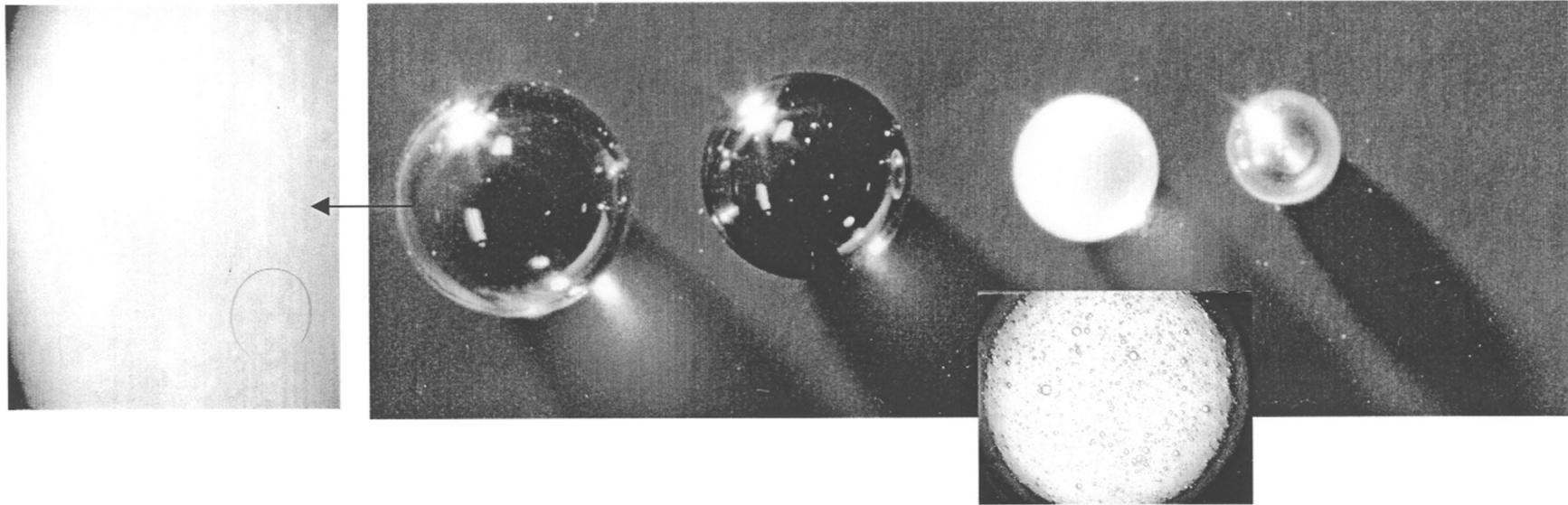


$\text{La}_3\text{Al}_5\text{O}_{12}$   $\text{LaErYAl}_5\text{O}_{12}$   $\text{Y}_3\text{Al}_5\text{O}_{12}$   $\text{Er}_3\text{Al}_5\text{O}_{12}$



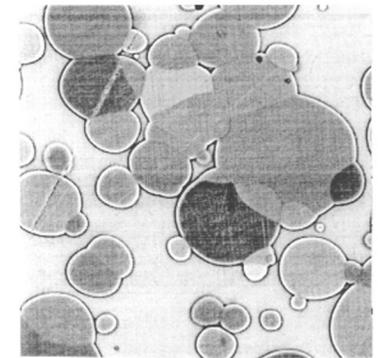
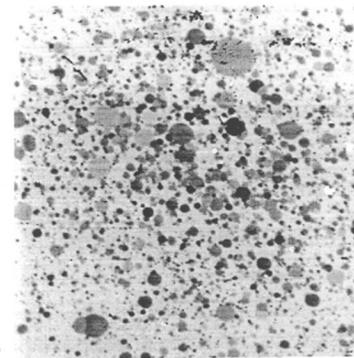
BSE YAG-alumina eutectic  
by Greg Jerman, NASA MSFC

# Glasses made from RE-Al oxide materials



WDS analysis results, average of 10 points

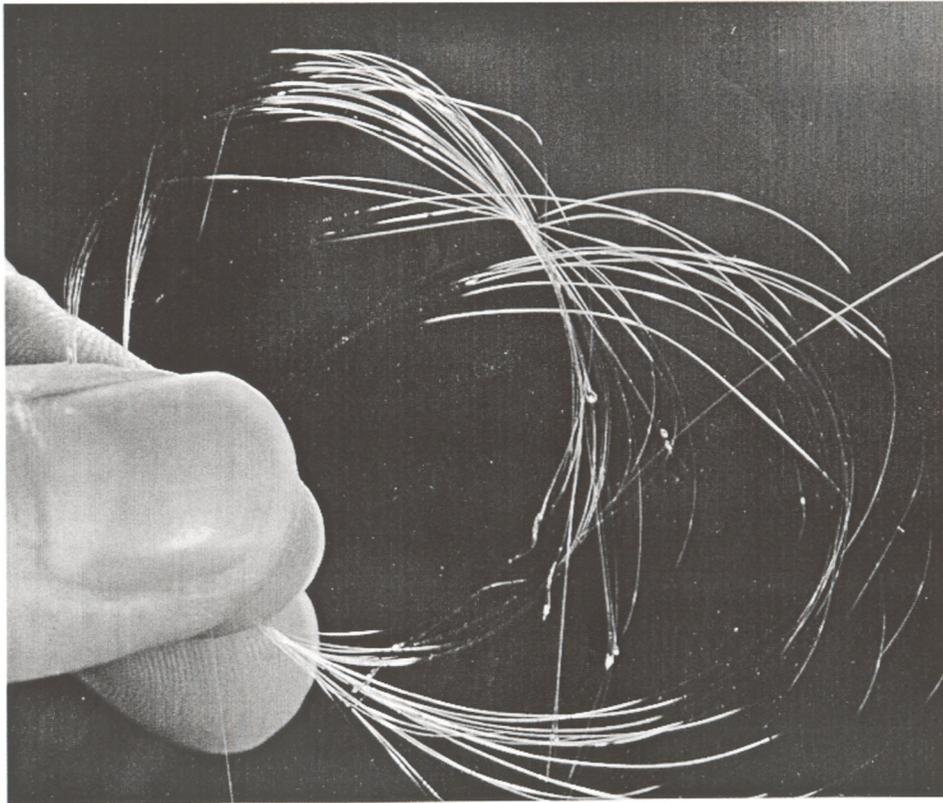
Area of sample	O (At%)	SD	Al (At%)	SD	Y (At%)	SD
Matrix	58.91	0.12	26.34	0.08	14.75	0.09
Light spheroids	58.70	0.10	26.45	0.06	14.86	0.05
Dark spheroids	58.61	0.09	26.55	0.06	14.84	0.09



[J. Am. Ceram. Soc., **83**, 1868 (2000)]

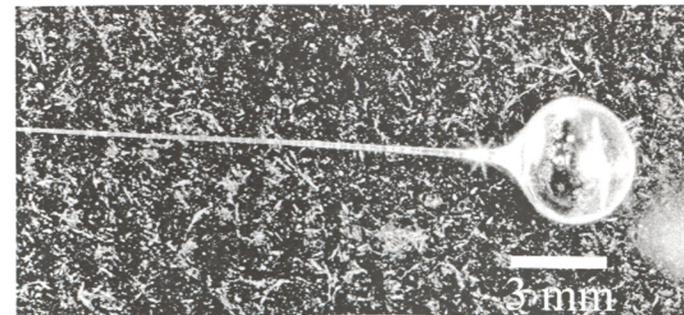
BSE

# $Y_3Al_5O_{12}$ -composition Glass Fibers

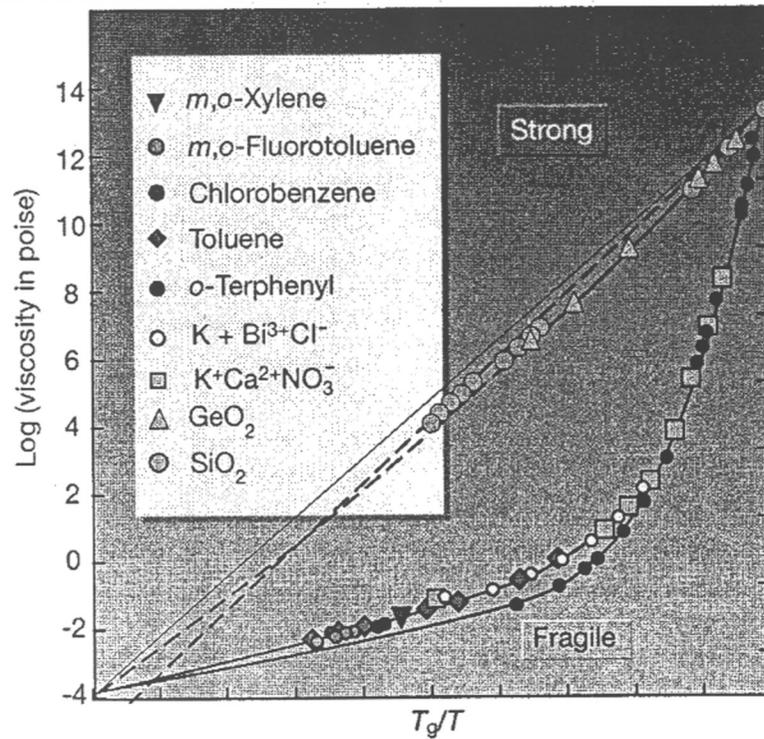


[Nature, 393, 769 (1998)]

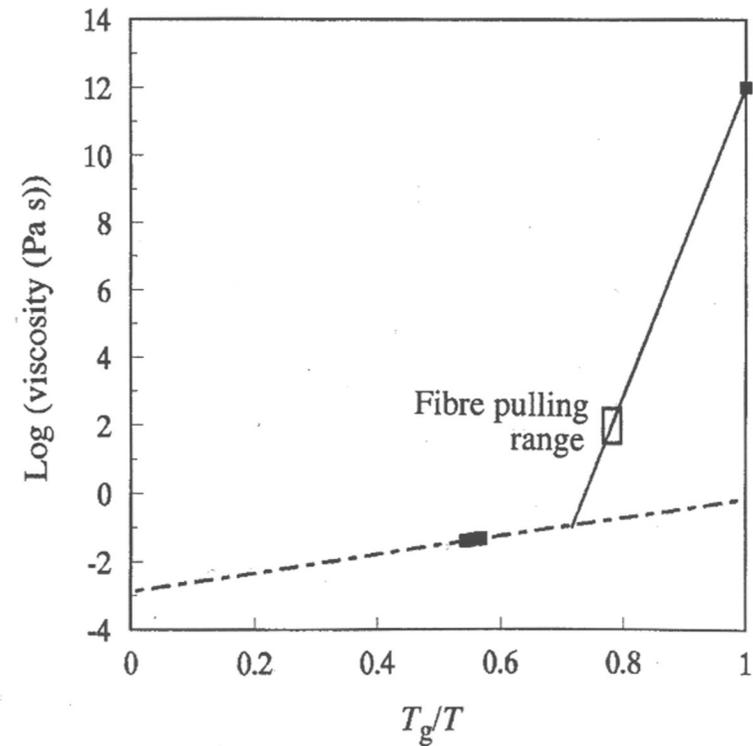
- Glass fibers pulled from liquid at 1600-1650 K (600 below  $T_m$ )
- Er-doped glass spheroid showing attenuation of a fiber.



# Strong and Fragile Liquids



[Adapted from Angell,  
Science, **267**, 1924 (1995)]

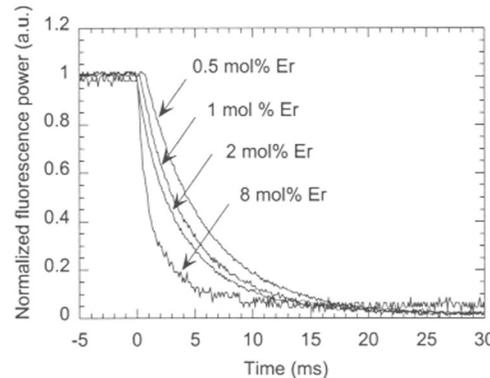
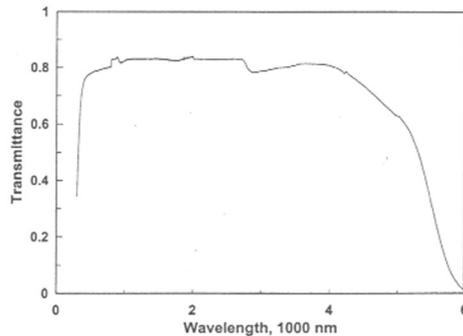


[Nature, **393**, 769 (1998)]

Fragility index,  $2(T_g/T_{1/2}) - 1$ , of YAG is about 0.65

# Some properties of REAl<sup>TM</sup> glass materials

- \* REAl glasses are: hard (900 Hv), strong, high refractive index (1.7), thermally ( $T_g > 1100$  K), chemically and mechanically stable, good solvents for RE ions, homogeneous dopant distribution
- \* One of the first “new” oxide glasses in about 30 years -- patents pending
- \* Applications include high power density lasers for *communications, surgery and rangefinding/detection*

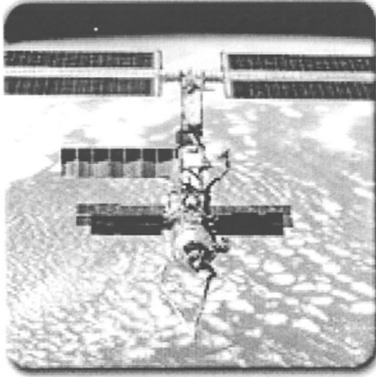


[J. Am. Ceram. Soc.,  
85, 1309 (2002)]

${}^4I_{13/2} > {}^4I_{15/2}$  transition in  $\text{Er}^{3+}$  ions

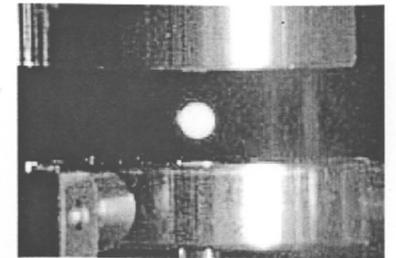
# Experimental Hypotheses

- A liquid-liquid phase transition can occur in undercooled A-Y melts by a diffusionless process
- Onset of the liquid-liquid phase transition is accompanied by a large change in the temperature dependence of viscosity

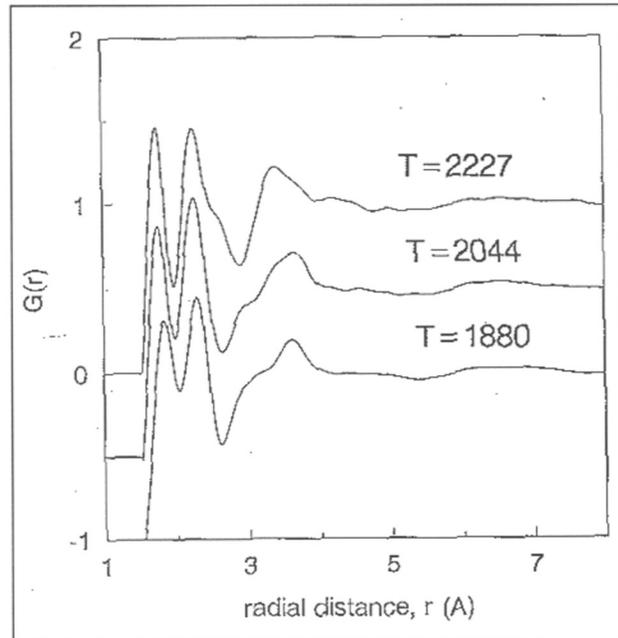


# Role of low gravity experiments

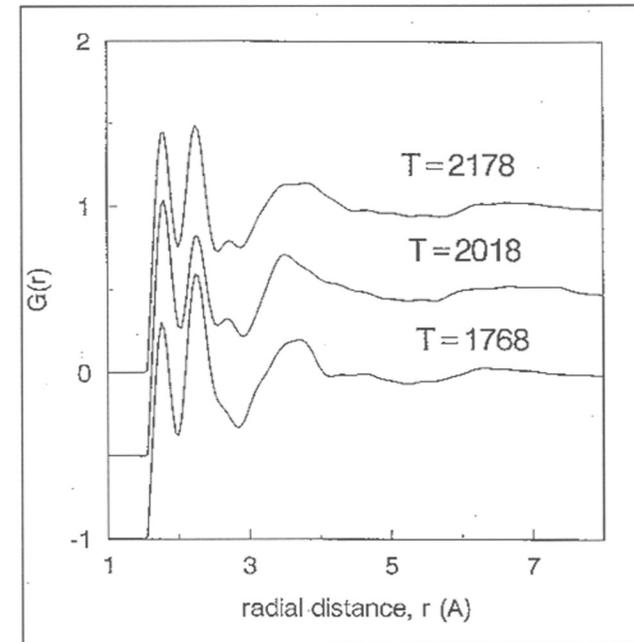
- We are studying kinetics in a liquid-liquid system: *it is essential to decouple diffusion and convection transport*
- The two phases have different densities: *sedimentation occurs on earth unless the melt is well stirred*
- Accurate measurement of melt viscosity requires a large and “quiescent” sample: *positioning forces can be minimized in low gravity allowing large samples to be used*



# Structure of molten $Y_3Al_5O_{12}$



Argon



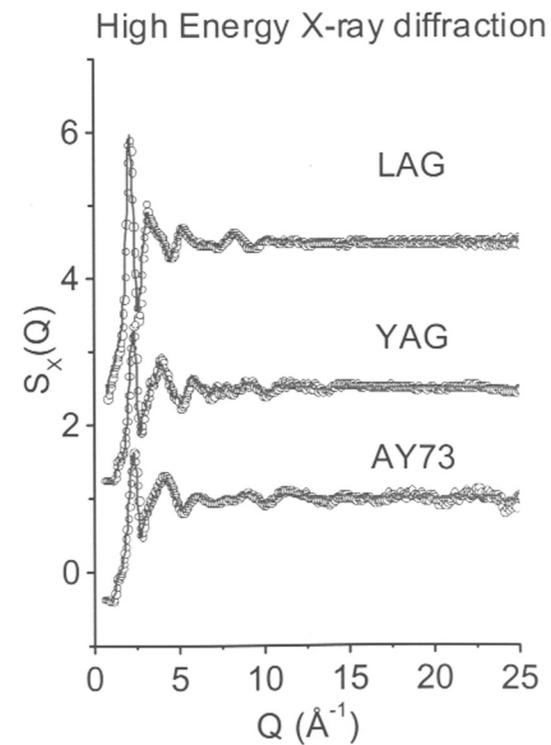
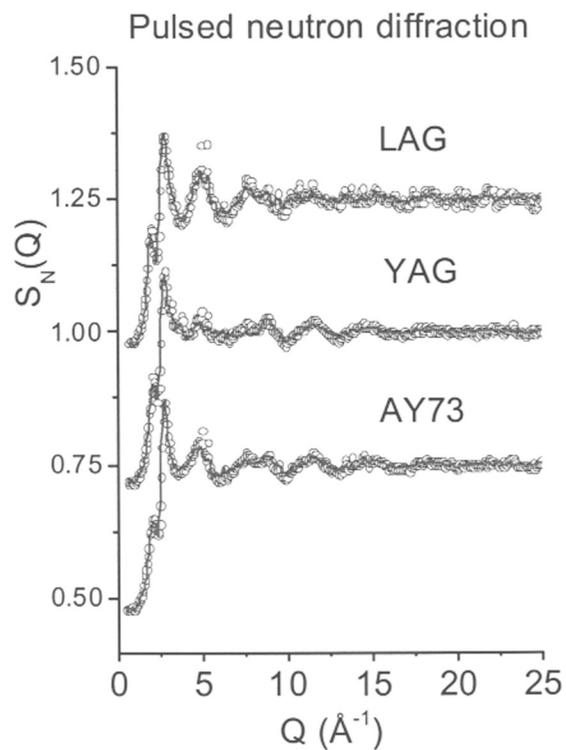
Oxygen

[Phys. Rev. Lett., **84**, 3622 (2000)]

Mainly 4CN Al-O and 6CN Y-O, changes with undercooling. In argon, which favors the L-L also increases population of 6 CN Al.

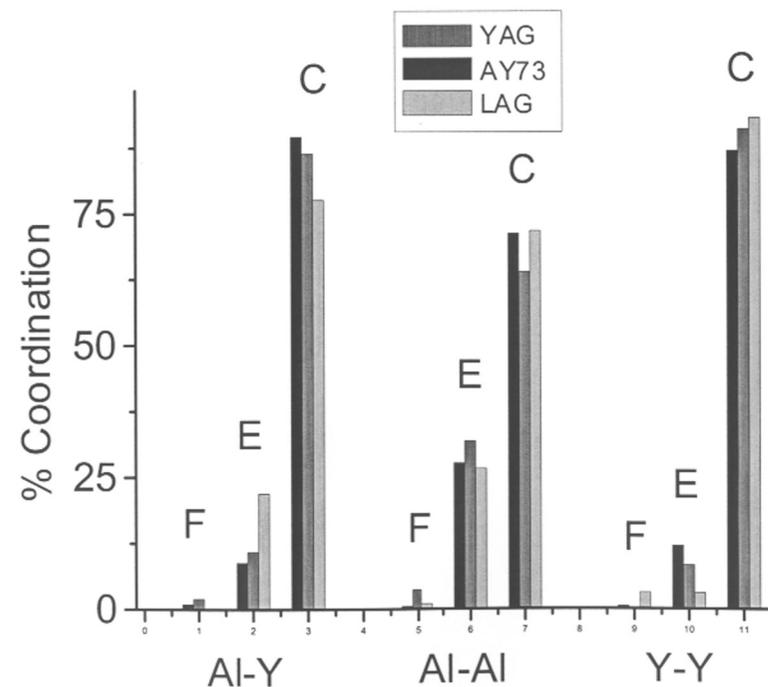
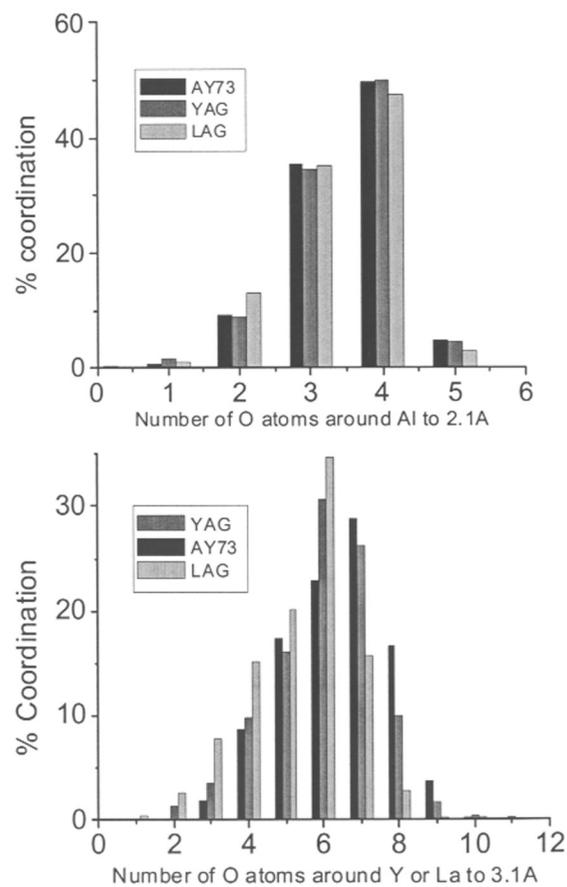
Is  $AlO_4 > AlO_6$  important in the L-L transition?

# Pulsed Neutron and H-E X-ray Structure Factors and RMC Simulation



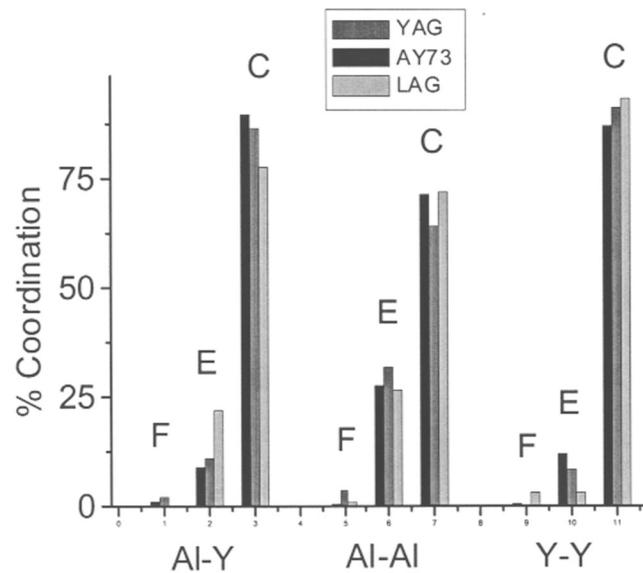
[Phys. Rev. Lett., submitted]

# Connectivity of M-O species and maintenance of M-O stoichiometry



[Phys. Rev. Lett., submitted]

# Connectivity of M-O species and maintenance of M-O stoichiometry



**O<sup>2-</sup>-generating reactions:**

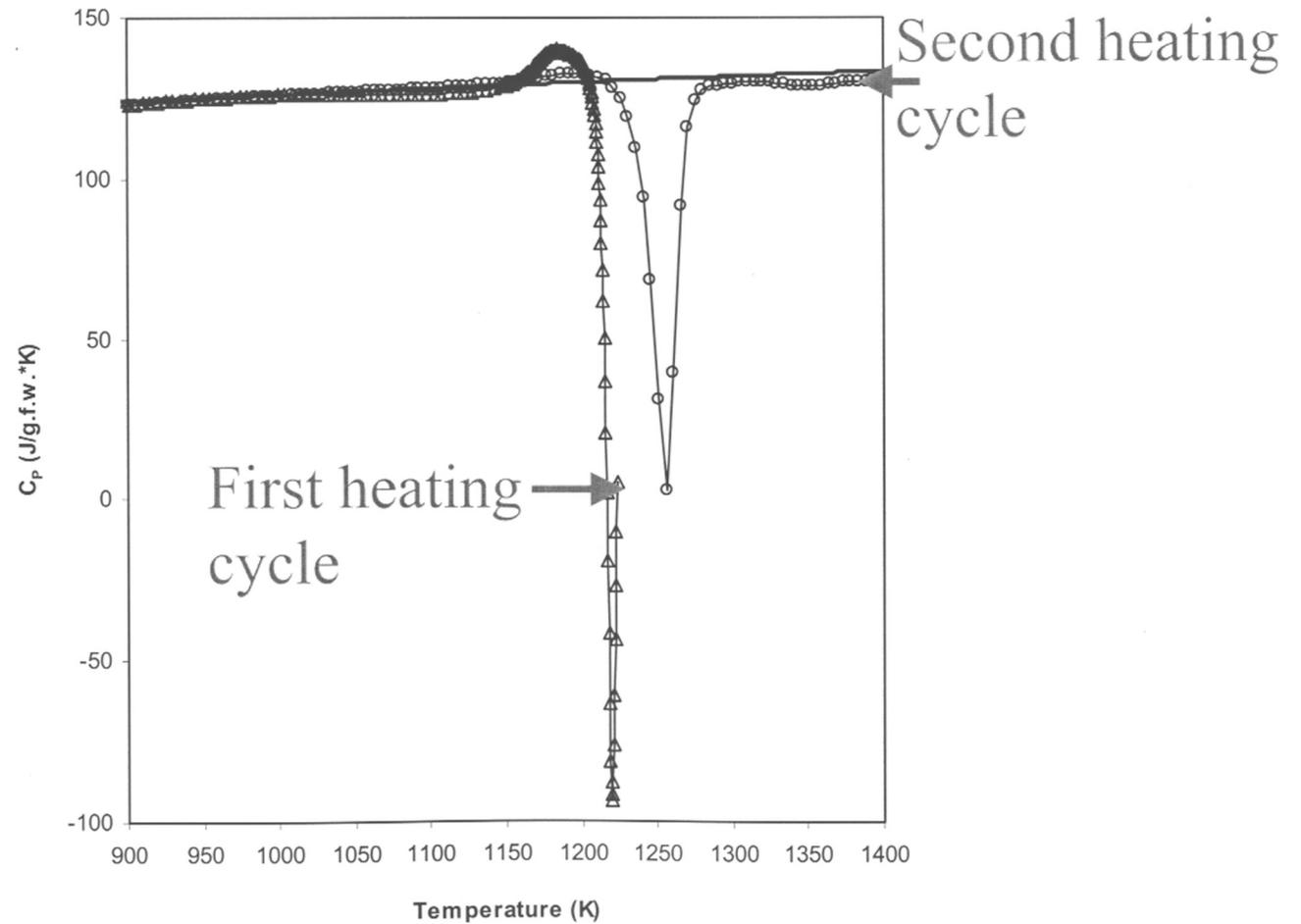
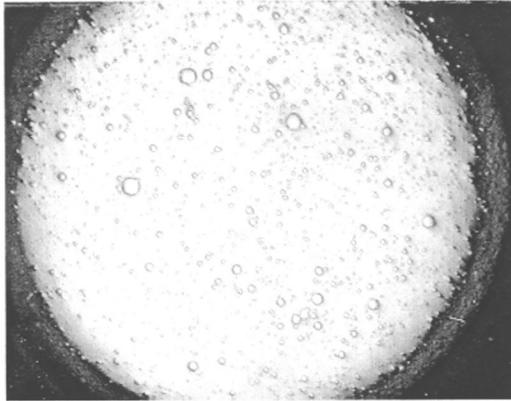


**O<sup>2-</sup>-consuming reaction:**

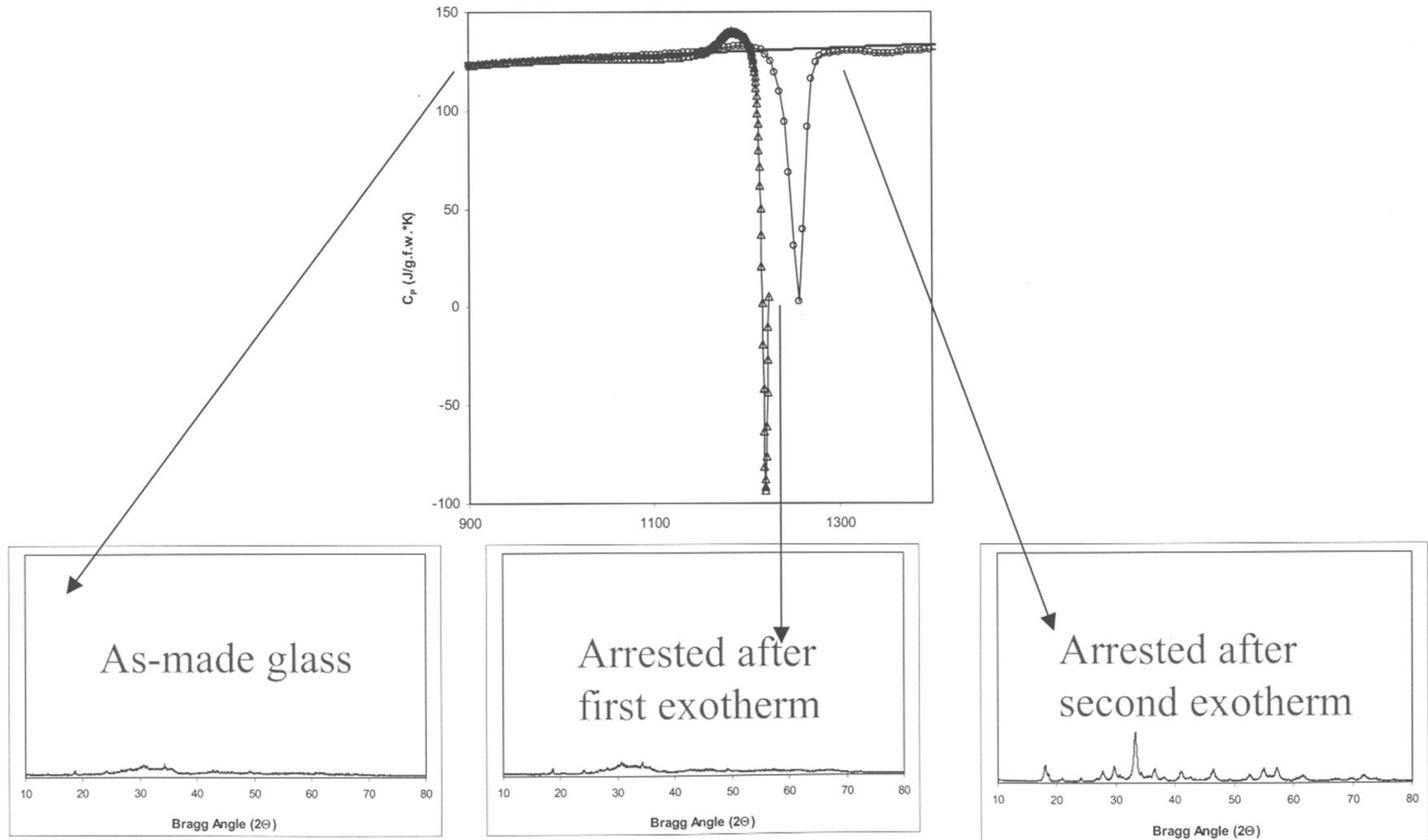


There is about 15% edge-shared YO species, after these are consumed, oxygen must be rejected if the  $\text{YO}_7 > \text{YO}_6$  reaction proceeds  
 - *how does this affect the phase transformation kinetics?*

# DSC on two-phase Y + 76 A-composition glass

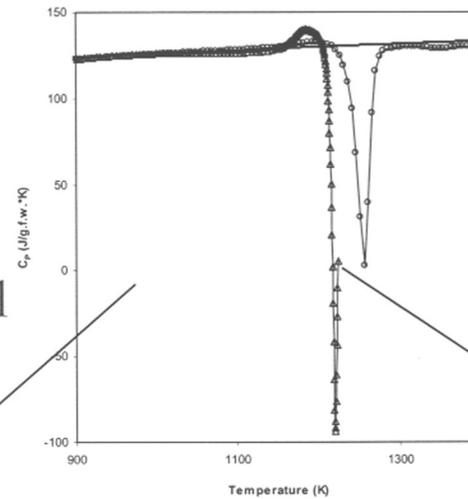
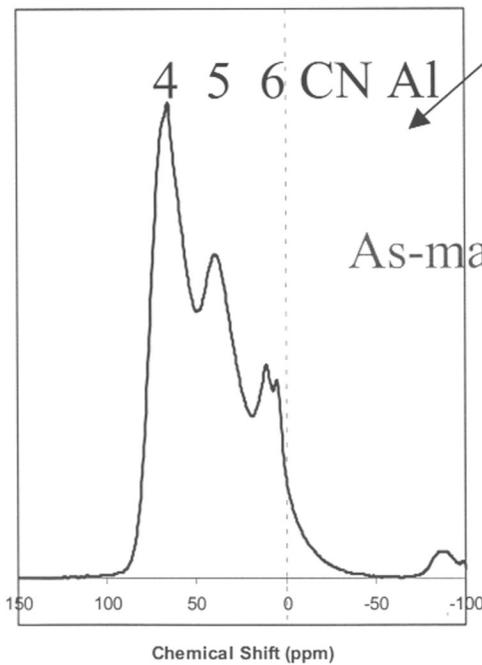


# XRD on as-made and heat treated Y + 76 A-composition glasses

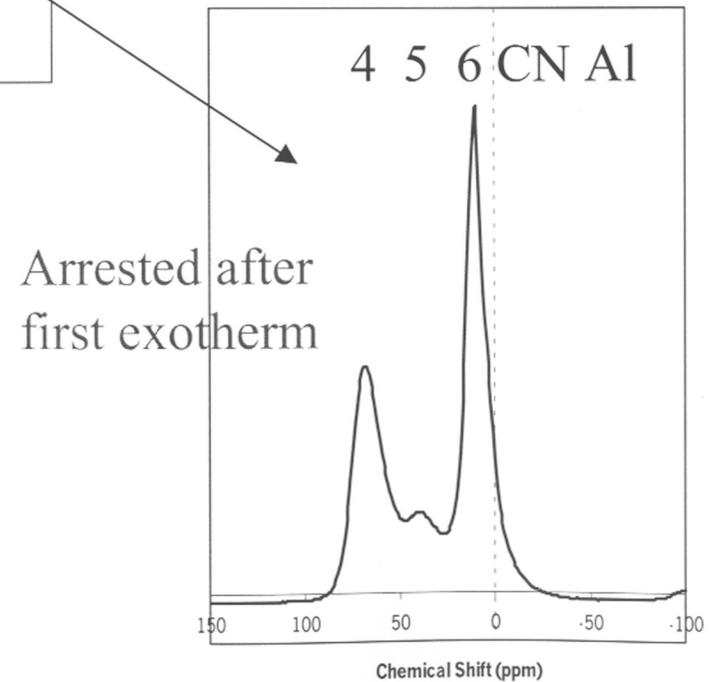


# High-Field $^{27}\text{Al}$ NMR on as-made and heat treated Y + 76 A-composition glasses

Analysis shows about 75/20/5% 4-,5- and 6-CN Al



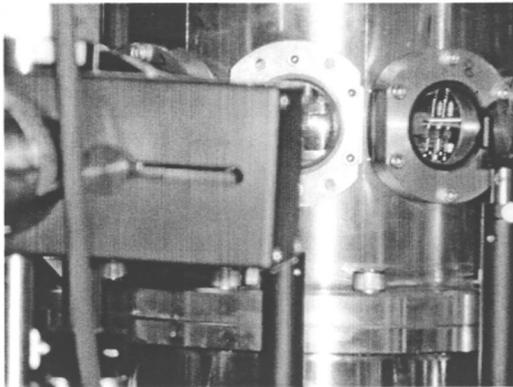
Analysis, microstructure, X-ray and neutron studies in progress....



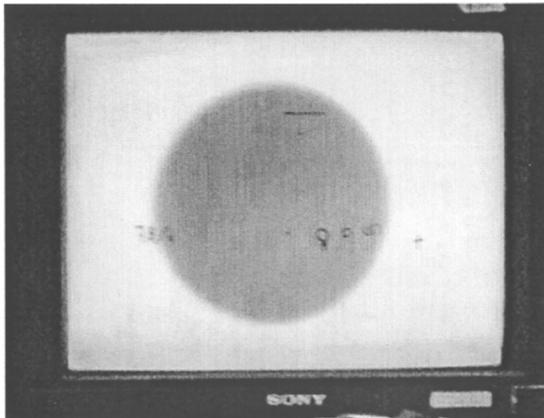
# Summary of structure results

- \* Transition involves both changes in CN and connectivity of polyhedra
- \* Stoichiometry can be preserved by edge- to corner-shared transformation/s to consume  $O^{2-}$
- \* Changes in Al CN are seen in NMR results
- \* Structure models are being optimized

# Measurements of melt properties



[RSI, 72, 2811 (2001), Jap. J. Appl. Phys., 41, 3030 (2002)]

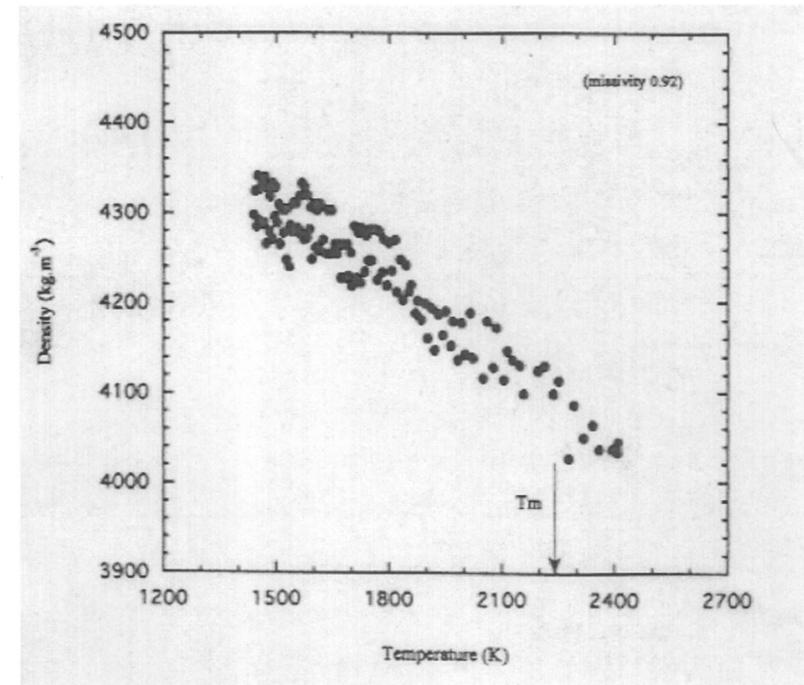


UV-backlit molten YAG sample

Density

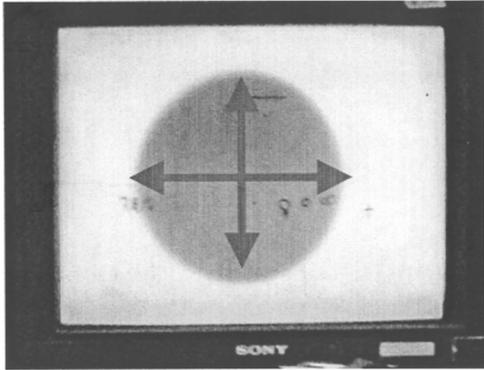
$$\rho = \frac{m}{4/3\pi r^3}$$

Preliminary measurements of the density of YAG(l) in pressurized ESL



[P-F. Paradis, *et al*, in prep.]

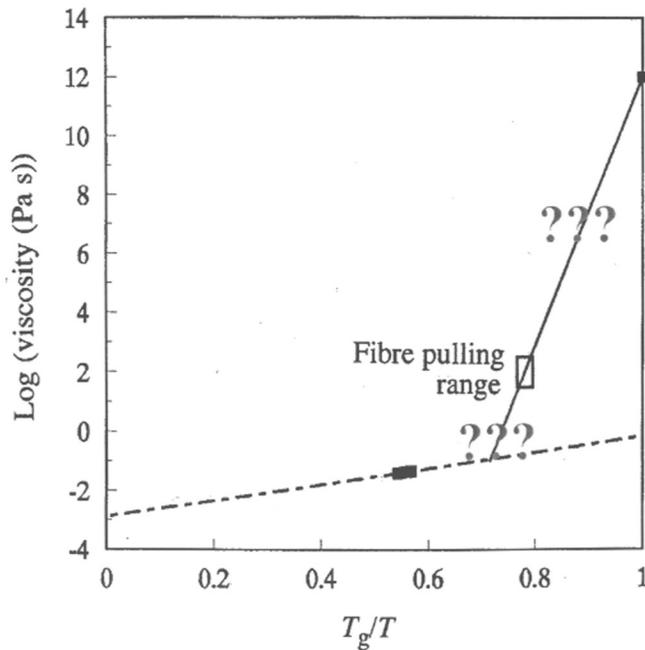
# Measurements of melt properties



Surface tension

$$\omega_l^2 = \frac{l(l-1)(l+2)\gamma}{\rho R_o^3}$$

Viscosity  $\tau_l = \frac{\rho R_o^2}{(l-1)(2l+1)\mu}$



$$\alpha^2 = \frac{\omega_R \rho R_o^2}{\mu} = \frac{\sqrt{l(l-1)(l+2)\gamma \rho R_o}}{\mu}$$

For  $\alpha^2 > 3.69$   
undamped

For damped ( $\alpha^2 < 3.69$ ) oscillations, measurements of shape relaxation *via* “polar altitude” can be used to determine viscosity. Very high viscosity region from creep if xtals do not form.

## Future/ongoing activities

- \* Investigation of transition kinetics by thermal processing of liquids and glasses
- \* Studies of liquid properties - *viscosity is a high priority* to establish correlation with str.
- \* Enhancement of structure models to enable a fuller understanding of CN and connectivity
- \* Establish detailed protocols/requirements for low gravity experiments

